

## THERMODYNAMIC VALUES FOR SUBSTANCES AT 298.15 K (25°C)

Substance	$\Delta H_f^\circ$ (kJ/mol)	$\Delta G_f^\circ$ (kJ/mol)	$S^\circ$ (J/mol-K)
<b>Aluminum</b>			
Al <sub>(s)</sub>	0	0	28.3
AlCl <sub>3(s)</sub>	-705.6	-630.0	109.3
Al <sub>2</sub> O <sub>3(s)</sub>	-1669.8	-1576.5	51.0
<b>Barium</b>			
Ba <sub>(s)</sub>	0	0	63.2
BaCO <sub>3(s)</sub>	-1216.3	-1137.6	112.1
BaO <sub>(s)</sub>	-553.5	-525.1	70.4
<b>Beryllium</b>			
Be <sub>(s)</sub>	0	0	9.44
BeO <sub>(s)</sub>	-608.4	-579.1	13.8
Be(OH) <sub>2(s)</sub>	-905.8	-817.9	50.2
<b>Bromine</b>			
Br <sub>(g)</sub>	111.8	82.4	174.9
Br <sup>-</sup> <sub>(aq)</sub>	-120.9	-102.8	80.7
Br <sub>2(g)</sub>	30.7	3.14	245.3
Br <sub>2(l)</sub>	0	0	152.3
HBr <sub>(g)</sub>	-36.2	-53.2	198.5
<b>Calcium</b>			
Ca <sub>(g)</sub>	179.3	145.5	154.8
Ca <sub>(s)</sub>	0	0	41.4
CaCO <sub>3(s, calcite)</sub>	-1207.1	1128.8	92.9
CaCl <sub>2(s)</sub>	-795.8	-748.1	104.6
CaF <sub>2(s)</sub>	-1219.6	-1167.3	68.9
CaO <sub>(s)</sub>	-635.5	-604.2	39.8
Ca(OH) <sub>2(s)</sub>	-986.2	-898.5	83.4
CaSO <sub>4(s)</sub>	-1434.0	-1321.8	106.7
<b>Carbon</b>			
C <sub>(g)</sub>	718.4	672.9	158.0

C(s, diamond)	1.88	2.84	2.43
C(s, graphite)	0	0	5.69
CCl <sub>4</sub> (g)	-106.7	-64.0	309.4
CCl <sub>4</sub> (l)	-139.3	-68.6	214.4
CF <sub>4</sub> (g)	-679.9	-635.1	262.3
CH <sub>4</sub> (g)	-74.8	-50.8	186.3
C <sub>2</sub> H <sub>2</sub> (g)	226.7	209.2	200.8
C <sub>2</sub> H <sub>4</sub> (g)	52.3	68.1	219.4
C <sub>2</sub> H <sub>6</sub> (g)	-84.7	-32.9	229.5
C <sub>3</sub> H <sub>8</sub> (g)	-103.9	-23.5	269.9
C <sub>4</sub> H <sub>10</sub> (g)	-124.7	-15.7	310.0
C <sub>4</sub> H <sub>10</sub> (l)	-147.6	-15.0	231.0
C <sub>6</sub> H <sub>6</sub> (g)	82.9	129.7	269.2
C <sub>6</sub> H <sub>6</sub> (l)	49.0	124.5	172.8
CH <sub>3</sub> OH(g)	-201.2	-161.9	237.6
CH <sub>3</sub> OH(l)	-238.6	-166.2	126.8
C <sub>2</sub> H <sub>5</sub> OH(g)	-235.1	-168.5	282.7
C <sub>2</sub> H <sub>5</sub> OH(l)	-277.7	-174.8	160.7
C <sub>6</sub> H <sub>12</sub> O <sub>6</sub> (s)	-1273.0	-910.4	212.1
CO(g)	-110.5	-137.2	197.9
CO <sub>2</sub> (g)	-393.5	-394.4	213.6
HC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> (l)	-487.0	-392.4	159.8
<b>Cesium</b>			
Cs(g)	76.5	49.5	175.6
Cs(l)	2.09	0.03	92.1
Cs(s)	0	0	85.2
CsCl(s)	-442.8	-414.4	101.2
<b>Chlorine</b>			
Cl(g)	121.7	105.7	165.2
Cl <sup>-</sup> (aq)	-167.2	-131.2	56.5
Cl <sub>2</sub> (g)	0	0	223.0
HCl(aq)	-167.2	-131.2	56.5
HCl(g)	-92.3	-95.3	186.7

<b>Chromium</b>			
Cr <sub>(g)</sub>	397.5	352.6	174.2
Cr <sub>(s)</sub>	0	0	23.6
Cr <sub>2</sub> O <sub>3(s)</sub>	-1139.7	-1058.1	81.2
<b>Cobalt</b>			
Co <sub>(g)</sub>	439	393	179
Co <sub>(s)</sub>	0	0	28.4
<b>Copper</b>			
Cu <sub>(g)</sub>	338.4	298.6	166.3
Cu <sub>(s)</sub>	0	0	33.3
CuCl <sub>2(s)</sub>	-205.9	-161.7	108.1
CuO <sub>(s)</sub>	-156.1	-128.3	42.6
Cu <sub>2</sub> O <sub>(s)</sub>	-170.7	-147.9	92.4
<b>Fluorine</b>			
F <sub>(g)</sub>	80.0	61.9	158.7
F <sub>(aq)</sub> <sup>-</sup>	-332.6	-278.8	-13.8
F <sub>2(g)</sub>	0	0	202.7
HF <sub>(g)</sub>	-268.6	-270.7	173.5
<b>Hydrogen</b>			
H <sub>(g)</sub>	217.9	203.3	114.6
H <sub>(aq)</sub> <sup>+</sup>	0	0	0
H <sub>(g)</sub> <sup>+</sup>	1536.2	1517.0	108.9
H <sub>2(g)</sub>	0	0	130.6
<b>Iodine</b>			
I <sub>(g)</sub>	106.6	70.2	180.7
I <sub>(aq)</sub> <sup>-</sup>	-55.2	-51.6	111.3
I <sub>2(g)</sub>	62.3	19.4	260.6
I <sub>2(s)</sub>	0	0	116.7
HI <sub>(g)</sub>	25.9	1.30	206.3
<b>Iron</b>			
Fe <sub>(g)</sub>	415.5	369.8	180.5
Fe <sub>(s)</sub>	0	0	27.2
Fe <sub>(aq)</sub> <sup>2+</sup>	-87.9	-84.9	113.4

$\text{Fe}^{3+}_{(\text{aq})}$	-47.7	-10.5	293.3
$\text{FeCl}_{2(\text{s})}$	-341.8	-302.3	117.9
$\text{FeCl}_{3(\text{s})}$	-400.	-334	142.3
$\text{FeO}_{(\text{s})}$	-271.9	-255.2	60.8
$\text{Fe}_2\text{O}_{3(\text{s})}$	-822.2	-741.0	90.0
$\text{Fe}_3\text{O}_{4(\text{s})}$	-1117.1	-1014.2	146.4
$\text{FeS}_{2(\text{s})}$	-171.5	-160.1	52.9
<b>Lead</b>			
$\text{Pb}_{(\text{s})}$	0	0	68.9
$\text{PbBr}_{2(\text{s})}$	-277.4	-260.7	161
$\text{PbCO}_{3(\text{s})}$	-699.1	-625.5	131.0
$\text{Pb}(\text{NO}_3)_{2(\text{aq})}$	-421.3	-246.9	303.3
$\text{Pb}(\text{NO}_3)_{2(\text{s})}$	-451.9	—	—
$\text{PbO}_{(\text{s})}$	-217.3	-187.9	68.7
<b>Lithium</b>			
$\text{Li}_{(\text{g})}$	159.3	126.6	138.8
$\text{Li}_{(\text{s})}$	0	0	29.1
$\text{Li}^{+}_{(\text{aq})}$	-278.5	-273.4	12.2
$\text{Li}^{+}_{(\text{g})}$	685.7	648.5	133.0
$\text{LiCl}_{(\text{s})}$	-408.3	-384.0	59.3
<b>Magnesium</b>			
$\text{Mg}_{(\text{g})}$	147.1	112.5	148.6
$\text{Mg}_{(\text{s})}$	0	0	32.5
$\text{MgCl}_{2(\text{s})}$	-641.6	-592.1	89.6
$\text{MgO}_{(\text{s})}$	-601.8	-569.6	26.8
$\text{Mg}(\text{OH})_{2(\text{s})}$	-924.7	-833.7	63.2
<b>Manganese</b>			
$\text{Mn}_{(\text{g})}$	280.7	238.5	173.6
$\text{Mn}_{(\text{s})}$	0	0	32.0
$\text{MnO}_{(\text{s})}$	-385.2	-362.9	59.7
$\text{MnO}_{2(\text{s})}$	-519.6	-464.8	53.1
$\text{MnO}_4^{-}_{(\text{aq})}$	-541.4	-447.2	191.2
<b>Mercury</b>			
$\text{Hg}_{(\text{g})}$	60.8	31.8	174.9

Hg(l)	0	0	77.4
HgCl <sub>2</sub> (s)	-230.1	-184.0	144.5
Hg <sub>2</sub> Cl <sub>2</sub> (s)	-264.9	-210.5	192.5
<b>Nickel</b>			
Ni(g)	429.7	384.5	182.1
Ni(s)	0	0	29.9
NiCl <sub>2</sub> (s)	-305.3	-259.0	97.65
NiO(s)	-239.7	-211.7	38.0
<b>Nitrogen</b>			
N(g)	472.7	455.5	153.3
N <sub>2</sub> (g)	0	0	191.5
NH <sub>3</sub> (aq)	-80.3	-26.5	111.3
NH <sub>3</sub> (g)	-46.2	-16.7	192.5
NH <sub>4</sub> <sup>+</sup> (aq)	-132.5	-79.3	113.4
N <sub>2</sub> H <sub>4</sub> (g)	95.4	159.4	238.5
NH <sub>4</sub> CN(s)	0	—	—
NH <sub>4</sub> Cl(s)	-314.4	-203.0	94.6
NH <sub>4</sub> NO <sub>3</sub> (s)	-365.6	-184.0	151
NO(g)	90.4	86.7	210.6
NO <sub>2</sub> (g)	33.8	51.8	240.5
N <sub>2</sub> O(g)	81.6	103.6	220.0
N <sub>2</sub> O <sub>4</sub> (g)	9.66	98.3	304.3
N <sub>2</sub> O <sub>5</sub> (g)	13.3	117.1	355.7
NOCl(g)	52.6	66.3	264
HNO <sub>3</sub> (aq)	-206.6	-110.5	146
HNO <sub>3</sub> (g)	-134.3	-73.9	266.4
<b>Oxygen</b>			
O(g)	247.5	230.1	161.0
O <sub>2</sub> (g)	0	0	205.0
O <sub>3</sub> (g)	142.3	163.4	237.6
OH <sup>-</sup> (aq)	-230.0	-157.3	-10.7
H <sub>2</sub> O(g)	-241.8	-228.6	188.8
H <sub>2</sub> O(l)	-285.8	-237.1	69.9
H <sub>2</sub> O <sub>2</sub> (g)	-136.1	-105.5	232.9

H <sub>2</sub> O <sub>2</sub> (l)	-187.8	-120.4	109.6
<b>Phosphorus</b>			
P(g)	316.4	280.0	163.2
P <sub>2</sub> (g)	144.3	103.7	218.1
P <sub>4</sub> (g)	58.9	24.4	280
P <sub>4</sub> (s, red)	-17.5	-12.0	22.9
P <sub>4</sub> (s, white)	0	0	41.1
PCl <sub>3</sub> (g)	-288.1	-269.6	311.7
PCl <sub>3</sub> (l)	-319.6	-272.4	217
PF <sub>5</sub> (g)	-1594.4	-1520.7	300.8
PH <sub>3</sub> (g)	5.4	13.4	210.2
P <sub>4</sub> O <sub>6</sub> (s)	-1640.1	—	—
P <sub>4</sub> O <sub>10</sub> (s)	-2940.1	-2675.2	228.9
POCl <sub>3</sub> (g)	-542.2	-502.5	325
POCl <sub>3</sub> (l)	-597.0	-520.9	222
H <sub>3</sub> PO <sub>4</sub> (aq)	-1288.3	-1142.6	158.2
<b>Potassium</b>			
K(g)	90.0	61.17	160.2
K(s)	0	0	64.7
KCl(s)	-435.9	-408.3	82.7
KClO <sub>3</sub> (s)	-391.2	-289.9	143.0
KClO <sub>3</sub> (aq)	-349.5	-284.9	265.7
K <sub>2</sub> CO <sub>3</sub> (s)	-1150.2	-1064.6	155.4
KNO <sub>3</sub> (s)	-492.7	-393.1	288.1
K <sub>2</sub> O(s)	-363.2	-322.1	94.1
KO <sub>2</sub> (s)	-284.5	-240.6	122.5
K <sub>2</sub> O <sub>2</sub> (s)	-495.8	-429.8	113.0
KOH(s)	-424.7	-378.9	78.91
KOH(aq)	-482.4	-440.5	91.6
<b>Rubidium</b>			
Rb(g)	85.8	55.8	170.0
Rb(s)	0	0	76.8
RbCl(s)	-430.5	-412.0	92

RbClO <sub>3(s)</sub>	-392.4	-292.0	152
<b>Scandium</b>			
Sc <sub>(g)</sub>	377.8	336.1	174.7
Sc <sub>(s)</sub>	0	0	34.6
<b>Selenium</b>			
H <sub>2</sub> Se <sub>(g)</sub>	29.7	15.9	219.0
<b>Silicon</b>			
Si <sub>(g)</sub>	368.2	323.9	167.8
Si <sub>(s)</sub>	0	0	18.7
SiC <sub>(s)</sub>	-73.2	-70.9	16.6
SiCl <sub>4(l)</sub>	-640.1	-572.8	239.3
SiO <sub>2(s, quartz)</sub>	-910.9	-856.5	41.8
<b>Silver</b>			
Ag <sub>(s)</sub>	0	0	42.6
Ag <sup>+</sup> <sub>(aq)</sub>	105.9	77.11	73.9
AgCl <sub>(s)</sub>	-127.0	-109.7	96.1
Ag <sub>2</sub> O <sub>(s)</sub>	-31.1	-11.2	121.3
AgNO <sub>3(s)</sub>	-124.4	-33.4	140.9
<b>Sodium</b>			
Na <sub>(g)</sub>	107.7	77.3	153.7
Na <sub>(s)</sub>	0	0	51.5
Na <sup>+</sup> <sub>(aq)</sub>	-240.1	-261.9	59.0
Na <sup>+</sup> <sub>(g)</sub>	609.3	574.3	148.0
NaBr <sub>(aq)</sub>	-360.6	-364.7	141.0
NaBr <sub>(s)</sub>	-361.4	-349.3	86.8
Na <sub>2</sub> CO <sub>3(s)</sub>	-1130.9	-1047.7	136.0
NaCl <sub>(aq)</sub>	-407.1	-393.0	115.5
NaCl <sub>(g)</sub>	-181.4	-201.3	229.8
NaCl <sub>(s)</sub>	-410.9	-384.0	72.3
NaHCO <sub>3(s)</sub>	-947.7	-851.8	102.1
NaNO <sub>3(aq)</sub>	-446.2	-372.4	207
NaNO <sub>3(s)</sub>	-467.9	-367.0	116.5

NaOH <sub>(aq)</sub>	-469.6	-419.2	49.8
NaOH <sub>(s)</sub>	-425.6	-379.5	64.5
<b>Strontium</b>			
SrO <sub>(s)</sub>	-592.0	-561.9	54.9
Sr <sub>(g)</sub>	164.4	110.0	164.6
<b>Sulfur</b>			
S <sub>(s, rhombic)</sub>	0	0	31.88
S <sub>8(g)</sub>	102.3	49.7	430.9
SO <sub>2(g)</sub>	-296.9	-300.4	248.5
SO <sub>3(g)</sub>	-395.2	-370.4	256.2
SO <sub>4<sup>2-</sup>(aq)</sub>	-909.3	-744.5	20.1
SOCl <sub>2(l)</sub>	-245.6	—	—
H <sub>2</sub> S <sub>(g)</sub>	-20.2	-33.0	205.6
H <sub>2</sub> SO <sub>4(aq)</sub>	-909.3	-744.5	20.1
H <sub>2</sub> SO <sub>4(l)</sub>	-814.0	-689.9	156.1
<b>Titanium</b>			
Ti <sub>(g)</sub>	468	422	180.3
Ti <sub>(s)</sub>	0	0	30.76
TiCl <sub>4(g)</sub>	-763.2	-726.8	354.9
TiCl <sub>4(l)</sub>	-804.2	-728.1	221.9
TiO <sub>2(s)</sub>	-944.7	-889.4	50.29
<b>Vanadium</b>			
V <sub>(g)</sub>	514.2	453.1	182.2
V <sub>(s)</sub>	0	0	28.9
<b>Zinc</b>			
Zn <sub>(g)</sub>	130.7	95.2	160.9
Zn <sub>(s)</sub>	0	0	41.63
ZnCl <sub>2(s)</sub>	-415.1	-369.4	111.5
ZnO <sub>(s)</sub>	-348.0	-318.2	43.9